



## Selenite foliar application increased the accumulation of medicinal components in *Paeonia ostii* by promoting antioxidant capacity, reducing oxidative stress, and improving photosynthetic capacity

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### Abstract

The effects of selenite (0, 15, 30, 45 mg L<sup>-1</sup>) on physiological characteristics and medicinal components of *Paeonia ostii* were analyzed. The results showed that selenite application promoted the activity of superoxide dismutase and the contents of soluble sugar, proline, carotenoids, total flavonoids, and total polyphenols, and decreased the contents of reactive oxygen species, relative electrical conductivity, and malondialdehyde. In addition, selenite also increased chlorophyll content, improved electron transfer ability, PSI and PSII performance, and the coordination between PSI and PSII, which significantly improved photosynthetic capacity. Moreover, selenite treatment also greatly increased the contents of gallic acid, catechin, albiflorin, paeoniflorin, benzoic acid, and paeonol in Moutan cortex radicis (MCR). These results showed that selenite effectively protected the photosynthetic apparatus from photooxidative damage by enhancing antioxidant capacity, improving photosynthetic capacity, and increasing the content of the medicinal compounds in MCR.

**Keywords:** chlorophyll fluorescence; *Paeonia ostii*; photosynthesis; secondary metabolites; selenite.

### Highlights

- Selenite improved the antioxidant system and reduced the oxidative stress in *Paeonia ostii*
- Selenite improved the performance of PSI and PSII, thereby promoting the increase of  $P_N$
- Selenite enhanced medicinal components in Moutan cortex radicis of *Paeonia ostii*

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**Abbreviations:** ABS/CS<sub>m</sub> – the absorbed energy flux per cross-section; ABS/RC – the absorption flux per reaction center; Car – carotenoid; Chl – chlorophyll; C<sub>i</sub> – intercellular carbon dioxide concentration; DI<sub>0</sub>/CS<sub>m</sub> – the dissipated energy flux per cross-section; DI<sub>0</sub>/RC – the dissipated energy flux per reaction center; E – transpiration rate; ET<sub>0</sub>/CS<sub>m</sub> – the electron transport flux per cross-section; ET<sub>0</sub>/RC – the electron transport flux per reaction center; ETR – the rate of electron transfer; F<sub>v</sub>/F<sub>m</sub> – the maximum photochemical efficiency of PSII under dark adaptation; F'<sub>v</sub>/F'<sub>m</sub> – the efficiency of excitation capture of open PSII center; g<sub>s</sub> – stomatal conductance; M<sub>0</sub> – the initial slope of the relative variable fluorescence of the relative rate at which Q<sub>A</sub> is reduced; MCR – Moutan cortex radicis; MDA – malondialdehyde; PI<sub>abs</sub> – the performance index on absorption basis; P<sub>N</sub> – the net photosynthetic rate; RC/CS<sub>m</sub> – the density of RCs per excited cross-section; REC – the relative electrical conductivity; ROS – reactive oxygen species; SOD – superoxide dismutase; TR<sub>0</sub>/CS<sub>m</sub> – the trapped energy flux per cross-section; TR<sub>0</sub>/RC – the trapped energy flux per reaction center; V<sub>J</sub> – the relative variable fluorescence intensity at the J step; W<sub>K</sub> – the normalized relative variable fluorescence; ΔI/I<sub>0</sub> – the maximum redox activity of PSI; φE<sub>0</sub> – the quantum yield for electron transport; Φ<sub>PSI/PSII</sub> – the coordination between PSI and PSII; Φ<sub>PSII</sub> – the actual photochemical efficiency of PSII.

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## Introduction

Tree peony (*Paeonia* sect. *Moutan*) is a famous traditional flower originating in China, which has not only ornamental value but also important medicinal value, and *Paeonia ostii* 'Fengdan' is one of the most representative cultivated varieties (Sun *et al.* 2021). Moutan cortex radicis (MCR) is the dry root bark of perennial tree peony root after removing the internal core, and it is rich in phenols, flavonoids, and terpenoids, such as paeonol, paeoniflorin, gallic acid, catechin, albiflorin, benzoic acid, *etc.* Studies have shown that these phenolic acids and flavonoids have antioxidant properties (Wang *et al.* 2019, Jucá *et al.* 2020), which are an important part of plant nonenzymatic antioxidants and another line of defense against reactive oxygen species (ROS) (Lanza and Reis 2021). MCR has a wide range of clinical applications, such as the functions of promoting blood circulation, cooling blood, clearing heat, and removing blood stasis, and is broadly used in traditional Chinese medicine for thousands of years (Wang *et al.* 2019). In China, MCR is used as an important crude drug in many traditional Chinese patent medicines to treat various diseases such as cardiovascular, extravasated blood, and tumors (Wang *et al.* 2017). Therefore, increasing the active components in MCR is of great importance to improve the quality of medicinal materials and the therapeutic effect on related diseases.

Studies have shown that the application of exogenous sodium selenite can regulate growth and development, as well as the synthesis and accumulation of secondary metabolites, thereby improving the intrinsic quality of crops (Li *et al.* 2022a). Selenium is an essential trace element for humans and plants, which is the constituent and active center of glutathione peroxidase (GSH-Px) (Puccinelli *et al.* 2020). Selenium is mainly absorbed by plants in the form of selenite (IV), selenate (VI), and organic selenium compounds; selenite and selenate are the two most widely used forms in selenium fertilizer applications. In plants, selenium in organic form is safer and more effective than selenium in inorganic form for human beings and animals (Rider *et al.* 2010, Kalaei *et al.* 2022). It is found that foliar application of selenite can promote the biosynthesis of organic selenium (IV) more effectively compared to selenate (VI) in onions, carrots (Kápolna *et al.* 2012), and corn (Longchamp *et al.* 2015). This view was also proved in wheat; wheat particles sprayed with selenite have a higher percentage of organic selenium than those sprayed with sodium selenate (Wang *et al.* 2020a). Therefore, the foliar application of sodium selenite is a popular selenium application method.

Selenium has a wide range of antioxidant capacities, which can increase the activities of GSH-Px, superoxide dismutase (SOD), peroxidase (POD), and other enzymes, thus increasing the scavenging capacity of ROS (Lanza and Reis 2021). The application of sodium selenite further increased the activities of antioxidant enzymes and alleviated the negative effects of cadmium stress in *Solanum lycopersicum* seedlings (Alyemeni *et al.* 2018). Quinoa plants showed a significant increase in SOD, catalase (CAT), POD, ascorbate peroxidase (APX), and

glutathione reductase (GR) activities, and a decrease in malondialdehyde (MDA) and hydrogen peroxide ( $H_2O_2$ ) at low concentrations of sodium selenite and sodium selenate (Khalofah *et al.* 2021). Selenium can also promote the increase of nonenzymatic active substances, soluble protein, soluble sugar, and proline content. In *Brassica pekinensis*, foliar spraying of sodium selenite significantly increased vitamin C, proline, protein, and sugar content (Liu *et al.* 2020). Similarly, spraying 10 mg L<sup>-1</sup> of sodium selenite significantly increased soluble protein and soluble sugar content in mustard (Li *et al.* 2023). In addition, selenium can regulate the synthesis of amino acids and proteins, as well as N-secondary compounds, such as phenolics and flavonoids with free radical scavenging activities (Malagoli *et al.* 2015). In mustard, the concentration of 10 mg L<sup>-1</sup> Na<sub>2</sub>SeO<sub>3</sub> significantly increased the contents of flavonoids, total phenols, vitamin C, and anthocyanins compared with the control (Li *et al.* 2023). Foliar spraying of sodium selenite under drought stress significantly increased total phenolic content, anthocyanins, and flavonoids in camelina and canola (Ahmad *et al.* 2021).

Photosynthesis consists of three important steps: primary reaction, including absorption, transfer and conversion of light energy, electron transport and photophosphorylation, and carbon assimilation step (Cruz and Avenson 2021). The normal operation of photosynthesis depends on the coordination of photosynthetic pigment, photosynthetic electron transport chain and PSI and PSII (Zhang *et al.* 2023). Recent studies showed that selenite played an effective role in promoting functionality of the photosynthetic apparatus. In *Billbergia zebrina*, selenite application improved the potential capacity of energy conservation in the photosynthetic apparatus and the electron transport dynamics between the intersystem and PSI (Souza *et al.* 2019). In *Medicago sativa*, selenite treatment greatly increased the pigment contents, and the maximal quantum yield of PSII photochemistry ( $F_v/F_m$ ) (Wang *et al.* 2022).

To the best of our knowledge, the literature lacks information on exogenous selenite application on the physiology and secondary metabolism accumulation, especially, its effects on PSI and PSII in tree peony. Therefore, it is necessary to study the effects of selenite on antioxidant activity, pigment content, photosynthetic parameters, structure and function of PSII and PSI, and medicinal component accumulation of tree peony. The aim of this study was to investigate whether there was a positive effect of selenite on *P. ostii* plants and whether the contents of beneficial phytochemicals (such as paeonol and paeoniflorin) increased in MCR.

## Materials and methods

**Plant material and experimental conditions:** Pot experiments were conducted in the experimental farm of Henan University of Science and Technology, Henan Province, China. In October 2021, healthy and uniform five-year-old *P. ostii* cv. 'Fengdan' seedlings were planted in plastic pots filled with 12 kg of mixed nutrient soil

( $V_{\text{soil}}:V_{\text{sand}}:V_{\text{organic matter}} = 3:1:1$ ), and then cultivated outside. The flower buds of *P. ostii* were removed in March 2022. Since 30 March, all seedlings were sprayed with 0.2%  $\text{KH}_2\text{PO}_4$  solution every 20 d for three times. A completely randomized design was used with 12 replications (pots) for each treatment. The seedlings were sprayed with sodium selenite solution (0, 15, 30, 45 mg  $\text{L}^{-1}$ ) for three times on 10 April, 30 April, and 20 May 2022, respectively. Using *Tween-80* (0.1%) as surfactant, selenite solution was sprayed on the adaxial and abaxial surfaces. The pots were covered with plastic film before spraying in case the selenite solution was introduced into the soil. All pots were watered normally to avoid drought stress. After 10 d of treatment, the physiological parameters were determined. MCR was harvested to determine the content of secondary metabolites in October 2022.

**Superoxide dismutase activity:** SOD (EC 1.15.1.1) activity was evaluated using the photochemical nitroblue tetrazolium (NBT) method described by Li *et al.* (2018). Fresh leaves (0.50 g) were homogenized in 50 mM phosphate buffered-saline (PBS, pH 7.8), and then centrifuged at 4,000 rpm at 4°C for 20 min to obtain the supernatant. Supernatant (0.2 mL) was mixed with 50 mM PBS, 100 mM ethylenediamine tetraacetic acid disodium salt, 130 mM methionine, 0.75 mM NBT, and 20  $\mu\text{M}$  riboflavin to form 3 mL of the reaction solution. The absorbance of the mixture at 560 nm was measured by a spectrophotometer (752N, INESA, China).

**Soluble protein, soluble sugar, and proline content:** The content of soluble protein was determined as described by Wang *et al.* (2015). First, 0.25 g of fresh leaves were homogenized in 10 mL of distilled water, then the supernatant was obtained after centrifugation at 3,000 rpm for 10 min at 4°C. Finally, the supernatant was mixed with 5.0 mL of *Coomassie Brilliant Blue G250* for 2 min, then the soluble protein content was calculated using the absorbance at 595 nm. The soluble protein content was expressed as  $\text{mg g}^{-1}$ .

The content of soluble sugar was determined as proposed by Dong *et al.* (2019). Fresh leaves (0.10 g) were placed in a test tube filled with distilled water, and extracted twice in boiling water for 30 min. Then, the extract (0.5 mL) was mixed with deionized water (1.5 mL), anthrone ethyl acetate reagent (0.5 mL), and concentrated sulfuric acid (5 mL), and boiled for 1 min. The absorbance of the extract at 630 nm was measured by a spectrophotometer (752N, INESA, China). The soluble sugar content was expressed as  $\text{mg g}^{-1}$ .

The proline content was determined according to Bates *et al.* (1973). Fresh leaves (0.20 g) were placed in a test tube, and 3% sulfosalicylic acid aqueous solution was added and extracted in boiling water for 10 min. Then 2 mL of supernatant was mixed with equal volumes of acetic acid and acidic ninhydrin and boiled for 0.5 h. Then toluene was added and fully shaken to obtain an extract. The absorbance was measured at 520 nm using a spectrophotometer (752N, INESA, China). The proline content was expressed as  $\mu\text{g g}^{-1}$ .

**Relative electrical conductivity and malondialdehyde content:** The relative electrical conductivity (REC) was assessed as previously described (Zhang *et al.* 2021). Fresh leaves (0.10 g) were rinsed and cut into thin strips, then mixed with 6 mL of distilled water, and incubated at room temperature for 24 h. The initial electrical conductivity (EC1) was measured with an electrical conductivity meter (DDS-11A, Shanghai, China). Then leaf samples were boiled for 0.5 h, and the electrical conductivity was determined again (EC2) after cooling, REC was calculated as the percentage ratio of  $\text{EC1/EC2} \times 100\%$ .

MDA was measured as described previously (Cai *et al.* 2019). Fresh leaves (0.25 g) were triturated using 5 mL of 5% (w/v) trichloroacetic acid and centrifuged at 10,000 rpm for 10 min. The supernatant (2 mL) was mixed with 2 mL (0.67%, w/v) thiobarbituric acid, then boiled for 30 min, cooled and centrifuged at 10,000 rpm for 10 min again, and then the absorbance was measured at 450, 532, and 600 nm. The MDA content was expressed as  $\text{nmol g}^{-1}$ .

**Determination of  $\text{H}_2\text{O}_2$  and superoxide radical ( $\text{O}_2^-$ ):**  $\text{H}_2\text{O}_2$  content was determined according to Cheeseman (2006). Fresh leaf samples (0.60 g) were ground in 1% trichloroacetic acid (5 mL) and centrifuged for 10 min at 4°C, then the supernatant (0.70 mL) was added to 10 mM PBS (0.7 mL, pH 7.0) and 1 M iodide potassium (1.4 mL). After 20 min in the dark, the absorbance of the supernatant at 390 nm was measured. The content of  $\text{H}_2\text{O}_2$  was expressed as  $\mu\text{mol g}^{-1}$ .

$\text{O}_2^-$  content was measured by the method of Zhang *et al.* (2021). Fresh leaf samples (1.0 g) were homogenized in 5 mL of PBS (65 mM, pH 7.8), filtered, and centrifuged for 10 min at 10,000 rpm. After that, the extract (2 mL) was mixed with 65 mM PBS (1.5 mL, pH 7.8) and 10 mM hydroxylamine hydrochloride (0.5 mL), and the mixture was incubated at 25°C for 20 min. The final mixture (2 mL) was transferred to a test tube containing 17 mM p-sulfanilic acid (2 mL) and 7 mM  $\alpha$ -naphthylamine (2 mL), incubated again at 30°C for 30 min. Finally, the  $\text{O}_2^-$  content was quantified spectrophotometrically at 530 nm. The content of  $\text{O}_2^-$  was expressed as  $\mu\text{g g}^{-1}$ .

**Photosynthetic pigments and gas-exchange parameters:** Pigment contents of chlorophyll (Chl) *a*, Chl *b*, and carotenoids (Car) were measured according to the method of Lichtenthaler (1987). Fresh leaves (0.10 g) were extracted with 80% acetone solution for 48 h in the dark. Then the absorbance of the extract was determined at 470, 646, and 663 nm. The pigment contents were expressed as  $\text{mg g}^{-1}$ .

At 9:00–11:00 h on sunny days, photosynthetic parameters, such as net photosynthetic rate ( $P_N$ ), intercellular  $\text{CO}_2$  concentration ( $C_i$ ), stomatal conductance ( $g_s$ ), and transpiration rate ( $E$ ), were measured using LI-6400 portable photosynthetic system (LI-COR, Lincoln, NE, USA). The detection conditions were as follows: illumination intensity was 1,200  $\mu\text{mol}(\text{photon}) \text{m}^{-2} \text{s}^{-1}$ , leaf chamber temperature was 25°C,  $\text{CO}_2$  concentration was 380  $\mu\text{mol mol}^{-1}$ , and the airflow rate was 500  $\mu\text{mol s}^{-1}$ .

Five biological repeats were performed for all treatments, with three leaves selected for each repetition.

**Chl fluorescence, the rapid Chl fluorescence induced curve (OJIP), and 820-nm modulation reflection curve:** Chl fluorescence was measured using a modulated fluorometer (*MINI-PAM 2000*, *WALZ*, Germany). The initial fluorescence in the light-adapted state ( $F_0'$ ) and maximal fluorescence in the light-adapted state ( $F_m'$ ) were measured by using saturating pulses and far-red light, respectively. The efficiency of excitation capture of open PSII center ( $F_v/F_m'$ ), the actual photochemical efficiency of PSII ( $\Phi_{PSII}$ ), and the rate of electron transfer (ETR) were then recorded according to the method of *Genty et al.* (1989).

The OJIP curve and 820-nm modulation reflection curve were simultaneously measured by *M-PEA* (*Hansatech*, Norfolk, UK) (*Strasser et al.* 2010). The leaves were dark-adapted for at least 1 h, then exposed to the saturated red light of  $3,000 \mu\text{mol}(\text{photon}) \text{m}^{-2} \text{s}^{-1}$  to obtain OJIP curve parameters (Appendix 1), the 820-nm modulated reflection curve was measured by exposure to  $250 \mu\text{mol}(\text{photon}) \text{m}^{-2} \text{s}^{-1}$  far-red light. The relative variable fluorescence ( $V_t$ ) at any time was calculated according to the formula:  $V_t = (F_t - F_0)/(F_m - F_0)$  (*Li et al.* 2020). The 820-nm modulation reflection curve was plotted according to the value of  $MR/MR_0$  ( $MR$  is the modulation reflection at different time points, and  $MR_0$  is the  $MR$  value of far-red light irradiated with 0.7 ms) (*Strasser et al.* 2010). The maximum redox activity of PSI ( $\Delta I/I_0$ ) and the coordination between PSI and PSII ( $\Phi_{PSI/PSII}$ ) were calculated according to the formula:  $\Delta I/I_0 = (I_0 - I_m)/I_m$ ,  $\Phi_{PSI/PSII} = (\Delta I/I_0)/\psi_0$  (*Zhang et al.* 2021). At least ten biological repeats were measured for each treatment.

**Total flavonoids, total phenolic, and secondary metabolites content:** The method proposed by *Chen et al.* (2019) was used to determine the content of total flavonoids. The MCR powder was placed in a conical flask filled with 60% ethanol solution, extracted in a water bath at  $70^\circ\text{C}$  for 1.5 h, and centrifuged at 4,000 rpm for 10 min. A volume of 0.25 mL of the above solution was added to 0.3 mL of 5%  $\text{NaNO}_2$  solution and shaken for 6 min. Then, 0.3 mL of 10%  $\text{Al}(\text{NO}_3)_3$  solution was added to the above mixture, shaken well, and left for 6 min, then 4 mL of  $\text{NaOH}$  (4%) solution was added, and the absorbance at 510 nm was measured after full shock and left for 10 min at constant volume. The content of total flavonoids was expressed as  $\text{mg g}^{-1}$ .

The total phenolic content was determined according to the method of *Feng et al.* (2014). MCR (0.2 g) was added to 5 mL of 80% methanol and extracted by ultrasound at  $25^\circ\text{C}$  for 5 min, centrifuged at 10,000 rpm for 10 min to obtain supernatant. After that, 5 mL of 80% methanol was added to the precipitation, and the supernatant was combined after repeated extraction for four times. Then, 0.1 mL of extract and 0.4 mL of distilled water were mixed with 2.5 mL of 10% Folin–Ciocalteu's reagent, and then 2 mL of 7.5%  $\text{Na}_2\text{CO}_3$  solution was added and incubated in

the dark at room temperature for 1 h. The absorption value was determined at 765 nm using a spectrophotometer (752N, *INESA*, China). The total phenolic content was expressed as  $\text{mg g}^{-1}$ .

The contents of gallic acid, catechin, albiflorin, paeoniflorin, benzoic acid, and paeonol were determined by HPLC according to previously described protocols (*Yu et al.* 2006). HPLC system comprised of *Agilent 1260* liquid chromatography (*Agilent Technology Inc.*, Santa Clara, CA, USA), *Waters C18* column (4.6 mm  $\times$  250 mm, 5  $\mu\text{m}$ ). The test solution was filtered with a 0.22- $\mu\text{m}$  organic membrane filter and detected using a gradient elution at 230 nm. The mobile phase consisted of a mixture of acetonitrile (A) and 0.05% phosphoric acid (pH 2.7, B). The gradient program was set as follows: 0  $\sim$  5 min, 8%  $\sim$  12% A; 5  $\sim$  20 min, 12  $\sim$  20% A; 20  $\sim$  25 min, 20% A; 25  $\sim$  35 min, 20  $\sim$  45% A; 35  $\sim$  40 min, 45% A; 40  $\sim$  50 min, 45  $\sim$  8% A. The injection volume was 20  $\mu\text{L}$ , the flow rate was set at  $1.0 \text{ mL min}^{-1}$ , and the column temperature was  $30^\circ\text{C}$ . The results were expressed as  $\text{mg g}^{-1}$ .

**Statistical analysis:** The measured data were analyzed by one-way analysis of variance (ANOVA) and *Duncan's* test ( $P < 0.05$ ) with *SPSS 16.0* (*SPSS Inc.*, Chicago, IL, USA). The measurement of biochemical parameters was determined by at least three biological repetitions.

## Results

**Effect of selenite on biochemical parameters:** With the increase of selenite concentration, SOD activity, soluble protein, soluble sugar, and proline contents enhanced first and then decreased (Fig. 1). These four indexes in all selenite treatments were greatly higher than that in the control and reached the maximum value with the  $30 \text{ mg L}^{-1}$  selenite treatment.

REC of all selenite treatments was lower than that of the control (Fig. 1), with the lowest value and a significant difference under  $30 \text{ mg}(\text{selenite}) \text{ L}^{-1}$ . MDA contents of all selenite treatments were significantly lower than that of the control, with the largest decrease in the  $30 \text{ mg}(\text{selenite}) \text{ L}^{-1}$ . The contents of  $\text{H}_2\text{O}_2$  and  $\text{O}_2^-$  of  $15\text{--}30 \text{ mg L}^{-1}$  selenite treatments were significantly lower than those in the control, with a minimum value at the  $30 \text{ mg}(\text{selenite}) \text{ L}^{-1}$ .

**Pigment content and photosynthetic parameters:** The contents of Chl *a*, Chl *b*, Car, and Chl (*a+b*) in all treatments showed a trend of rising first and then decreasing (Fig. 2). Chl *a* content of all selenite treatments was obviously higher than that of control. The changes in Chl *b* and Car content were similar, which were higher than the control in all selenite treatments, and markedly higher than the control only at the  $30 \text{ mg}(\text{selenite}) \text{ L}^{-1}$ . The content of Chl (*a+b*) was significantly higher in all selenite treatments than that in the control. All pigment contents reached the maximum under the  $30 \text{ mg L}^{-1}$  selenite treatment.

The  $P_N$  under all selenite treatments was significantly higher than that of the control under  $15\text{--}30 \text{ mg L}^{-1}$  selenite

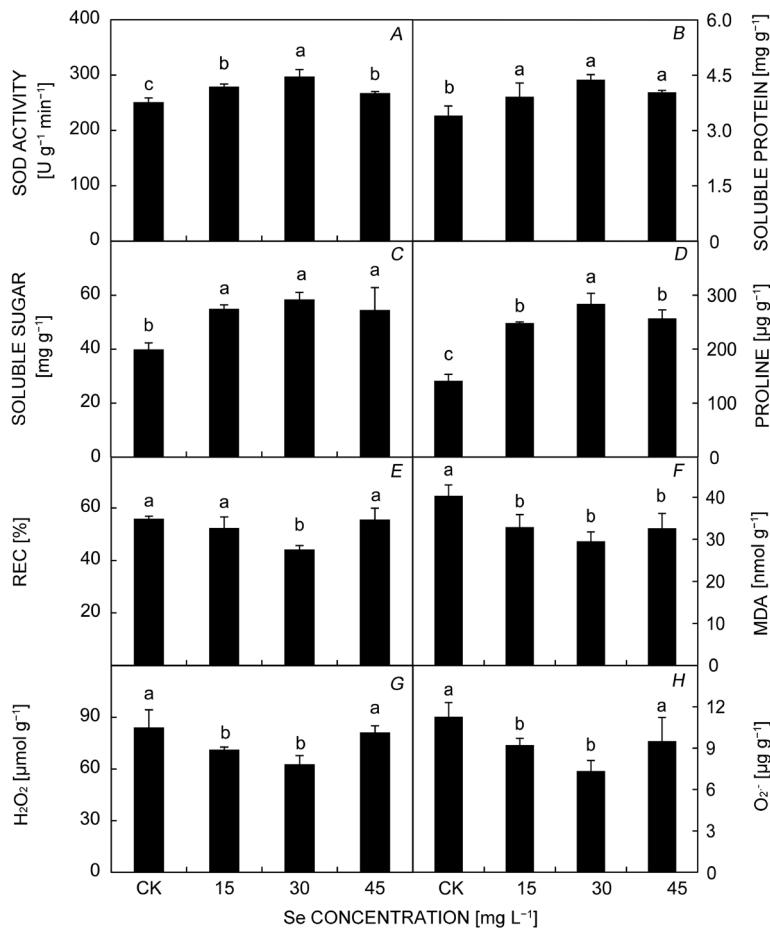


Fig. 1. Effect of different selenite treatments on the activity of SOD (A), and the content of soluble protein (B), soluble sugar (C), proline (D), REC (E), MDA (F),  $\text{H}_2\text{O}_2$  (G), and  $\text{O}_2^-$  (H) in *Paeonia ostii* leaves. Data represent the mean  $\pm$  SD,  $n = 3$ . SOD – superoxide dismutase; MDA – malondialdehyde; REC – relative electrical conductivity. Different lowercase letters above the bars indicate a significant difference between treatments at  $P < 0.05$  as determined by a Duncan's multiple range test.

treatment (Fig. 2). The change trends of  $g_s$  and  $E$  under different selenite treatments were similar, both increased significantly in all selenite treatments. Compared with the control group, the  $C_i$  of all selenite treatments increased, the increase was significant in 15 and 30  $\text{mg L}^{-1}$  selenite treatments.

**Effects of selenite on Chl fluorescence parameters:** The  $\text{F}_v'/\text{F}_m'$ ,  $\Phi_{\text{PSII}}$ , and ETR all increased first and then decreased with the increase of selenite concentration (Table 1). These three parameters all increased under all selenite treatments and reached the maximum value under the selenite treatment of 30  $\text{mg L}^{-1}$ , which was significantly higher than the control.

**Chl fluorescence OJIP curve and 820-nm modulation reflection curve:** As shown in Fig. 3A, the K point ( $t = 0.3$  ms) and J point ( $t = 2$  ms) decreased in all selenite treatments, and the OJIP curve changed most obviously under 30  $\text{mg L}^{-1}$  selenite treatment. The  $\text{MR}/\text{MR}_0$  of the 820-nm reflection fluorescence absorption curve (Fig. 3B) in *P. ostii* decreased rapidly from 0.7 ms of JIP time to 3 ms linearly, and then slowly decreased to the minimum value between 3 and 30 ms. The minimum  $\text{MR}/\text{MR}_0$  values of all selenite treatments decreased obviously, and 30  $\text{mg L}^{-1}$  selenite treatment had the largest decrease, followed by 15 and 45  $\text{mg L}^{-1}$  selenite treatments.

**Fast Chl fluorescence parameters, the function and coordination of PSII and PSI of *P. ostii* leaves:** The normalized fluorescence ( $W_K$ ) of the K phase and the relative fluorescence change ( $V_J$ ) of the J phase were calculated to quantify the changes of the K and J phase in the OJIP curve (Fig. 4). The values of  $W_K$ ,  $V_J$ , and the initial slope of the relative variable fluorescence of the relative rate at which  $Q_A$  is reduced ( $M_0$ ) decreased first and then increased, while the quantum yield for electron transport ( $\varphi E_0$ ) increased first and then decreased. The  $W_K$  and  $M_0$  in all selenite treatments were significantly lower than those of the control (Fig. 4). In terms of  $V_J$  parameters, only  $V_J$  in 30  $\text{mg L}^{-1}$  selenite treatment was significantly lower than that of the control. The  $\varphi E_0$  under 15–30  $\text{mg L}^{-1}$  selenite treatment was significantly higher than that of the control, except for the 45  $\text{mg L}^{-1}$  selenite treatment.

The parameters of the absorption flux per reaction center (ABS/RC), the dissipated energy flux per reaction center (DI<sub>0</sub>/RC), the trapped energy flux per reaction center (TR<sub>0</sub>/RC), and the electron transport flux per reaction center (ET<sub>0</sub>/RC) decreased at first and then increased with the increase of selenite concentration (Table 1). The four parameters in all selenite treatments were significantly lower than those in the control, and the minimum values were all obtained in the 30  $\text{mg L}^{-1}$  selenite treatment. The density of RCs per excited cross-section (RC/CS<sub>m</sub>), the absorbed energy flux per cross-section (ABS/CS<sub>m</sub>),

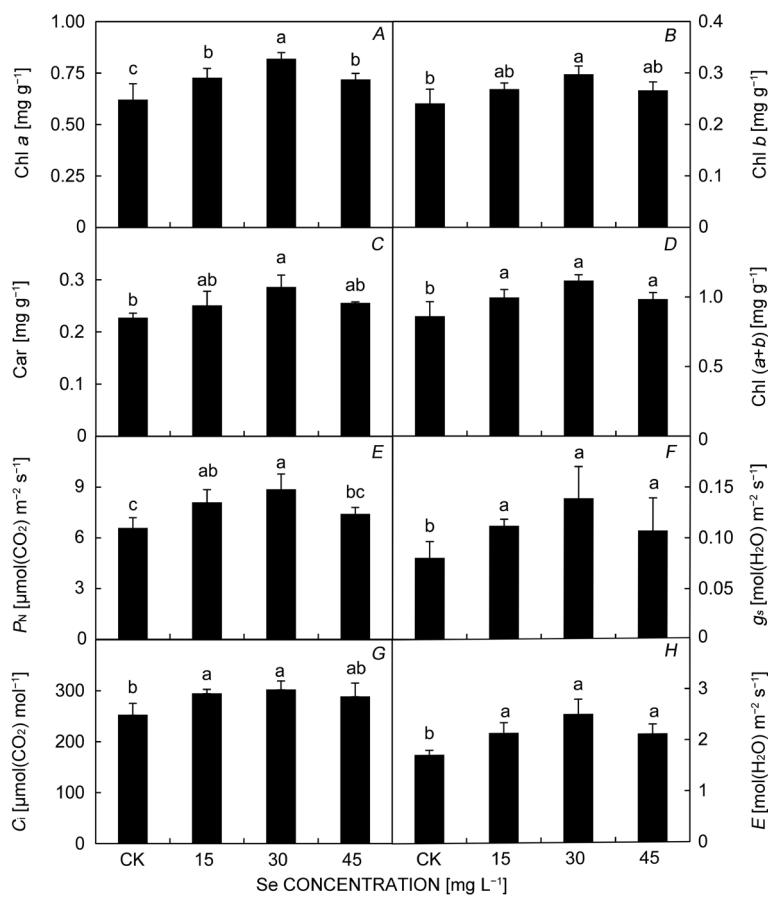


Fig. 2. Effect of different selenite treatments on the contents of Chl *a* (*A*), Chl *b* (*B*), Car (*C*), total Chl (*a+b*) (*D*), and *P<sub>n</sub>* (*E*), *g<sub>s</sub>* (*F*), *C<sub>i</sub>* (*G*), *E* (*H*) in *Paeonia ostii* leaves. Chl – chlorophyll; Car – carotenoid; *P<sub>n</sub>* – the net photosynthetic rate; *g<sub>s</sub>* – stomatal conductance; *C<sub>i</sub>* – intercellular carbon dioxide concentration; *E* – transpiration rate. Data represent the mean  $\pm$  SD, *n* = 3 for leaf photosynthetic pigment concentrations, *n* = 12 for the photosynthetic parameters. Different lowercase letters above the bars indicate a significant difference between treatments at *P* < 0.05 as determined by a *Duncan's* multiple range test.

Table 1. The changes of the fluorescence parameters and the energy fluxes in *Paeonia ostii* leaves under different selenite treatments [mg L⁻¹]. *F<sub>v</sub>'/F<sub>m</sub>'* – the efficiency of excitation capture of open PSII center;  $\Phi_{\text{PSII}}$  – the actual photochemical efficiency of PSII; ETR – the rate of electron transfer; ABS/RC – the absorption flux per reaction center; TR<sub>0</sub>/RC – the trapped energy flux per reaction center; ET<sub>0</sub>/RC – the electron transport flux per reaction center; DI<sub>0</sub>/RC – the dissipated energy flux per reaction center; RC/CS<sub>m</sub> – the density of RCs per excited cross-section; ABS/CS<sub>m</sub> – the absorbed energy flux per cross-section; TR<sub>0</sub>/CS<sub>m</sub> – the trapped energy flux per cross-section; ET<sub>0</sub>/CS<sub>m</sub> – the electron transport flux per cross-section; DI<sub>0</sub>/CS<sub>m</sub> – the dissipated energy flux per cross-section; F<sub>v</sub>/F<sub>m</sub> – the maximum photochemical efficiency of PSII under dark adaptation; PI<sub>abs</sub> – the performance index on absorption basis. Data represent the mean  $\pm$  SD (*n* = 12). Different lowercase letters in each column indicate a significant difference between treatments at *P* < 0.05 as determined by a *Duncan's* multiple range test.

Parameters	0 (CK)	15	30	45
<i>F<sub>v</sub>'/F<sub>m</sub>'</i>	0.58 $\pm$ 0.04 <sup>b</sup>	0.61 $\pm$ 0.02 <sup>b</sup>	0.67 $\pm$ 0.01 <sup>a</sup>	0.60 $\pm$ 0.01 <sup>b</sup>
$\Phi_{\text{PSII}}$	0.45 $\pm$ 0.03 <sup>b</sup>	0.48 $\pm$ 0.01 <sup>b</sup>	0.52 $\pm$ 0.01 <sup>a</sup>	0.47 $\pm$ 0.02 <sup>b</sup>
ETR	36.97 $\pm$ 2.50 <sup>b</sup>	39.13 $\pm$ 0.91 <sup>b</sup>	42.53 $\pm$ 0.93 <sup>a</sup>	39.07 $\pm$ 0.90 <sup>b</sup>
ABS/RC	1.42 $\pm$ 0.09 <sup>a</sup>	1.17 $\pm$ 0.03 <sup>b</sup>	1.03 $\pm$ 0.07 <sup>c</sup>	1.18 $\pm$ 0.03 <sup>b</sup>
DI <sub>0</sub> /RC	0.43 $\pm$ 0.05 <sup>a</sup>	0.29 $\pm$ 0.02 <sup>b</sup>	0.26 $\pm$ 0.04 <sup>b</sup>	0.31 $\pm$ 0.02 <sup>b</sup>
TR <sub>0</sub> /RC	1.02 $\pm$ 0.07 <sup>a</sup>	0.88 $\pm$ 0.02 <sup>b</sup>	0.78 $\pm$ 0.03 <sup>c</sup>	0.87 $\pm$ 0.05 <sup>b</sup>
ET <sub>0</sub> /RC	0.58 $\pm$ 0.04 <sup>a</sup>	0.52 $\pm$ 0.02 <sup>b</sup>	0.50 $\pm$ 0.02 <sup>b</sup>	0.51 $\pm$ 0.01 <sup>b</sup>
RC/CS <sub>m</sub>	13,431.37 $\pm$ 1,416.73 <sup>c</sup>	17,641.92 $\pm$ 895.12 <sup>b</sup>	21,261.11 $\pm$ 1,687.24 <sup>a</sup>	17,435.37 $\pm$ 542.11 <sup>b</sup>
ABS/CS <sub>m</sub>	23,803.00 $\pm$ 666.18 <sup>b</sup>	24,728.75 $\pm$ 796.92 <sup>ab</sup>	26,263.67 $\pm$ 762.27 <sup>a</sup>	24,654.33 $\pm$ 1,213.88 <sup>ab</sup>
DI <sub>0</sub> /CS <sub>m</sub>	7,251.00 $\pm$ 151.92 <sup>a</sup>	6,173.50 $\pm$ 233.26 <sup>b</sup>	6,326.33 $\pm$ 590.53 <sup>b</sup>	6,245.00 $\pm$ 209.82 <sup>b</sup>
TR <sub>0</sub> /CS <sub>m</sub>	16,985.33 $\pm$ 641.74 <sup>c</sup>	18,805.25 $\pm$ 645.88 <sup>ab</sup>	19,270.67 $\pm$ 628.02 <sup>a</sup>	17,742.67 $\pm$ 563.34 <sup>bc</sup>
ET <sub>0</sub> /CS <sub>m</sub>	10,446.00 $\pm$ 392.74 <sup>b</sup>	11,234.75 $\pm$ 683.91 <sup>b</sup>	12,433.33 $\pm$ 537.67 <sup>a</sup>	11,045.67 $\pm$ 397.05 <sup>b</sup>
F <sub>v</sub> /F <sub>m</sub>	0.71 $\pm$ 0.02 <sup>b</sup>	0.75 $\pm$ 0.01 <sup>a</sup>	0.76 $\pm$ 0.02 <sup>a</sup>	0.74 $\pm$ 0.03 <sup>ab</sup>
PI <sub>abs</sub>	2.35 $\pm$ 0.19 <sup>c</sup>	3.93 $\pm$ 0.72 <sup>b</sup>	5.43 $\pm$ 0.85 <sup>a</sup>	3.43 $\pm$ 0.28 <sup>b</sup>

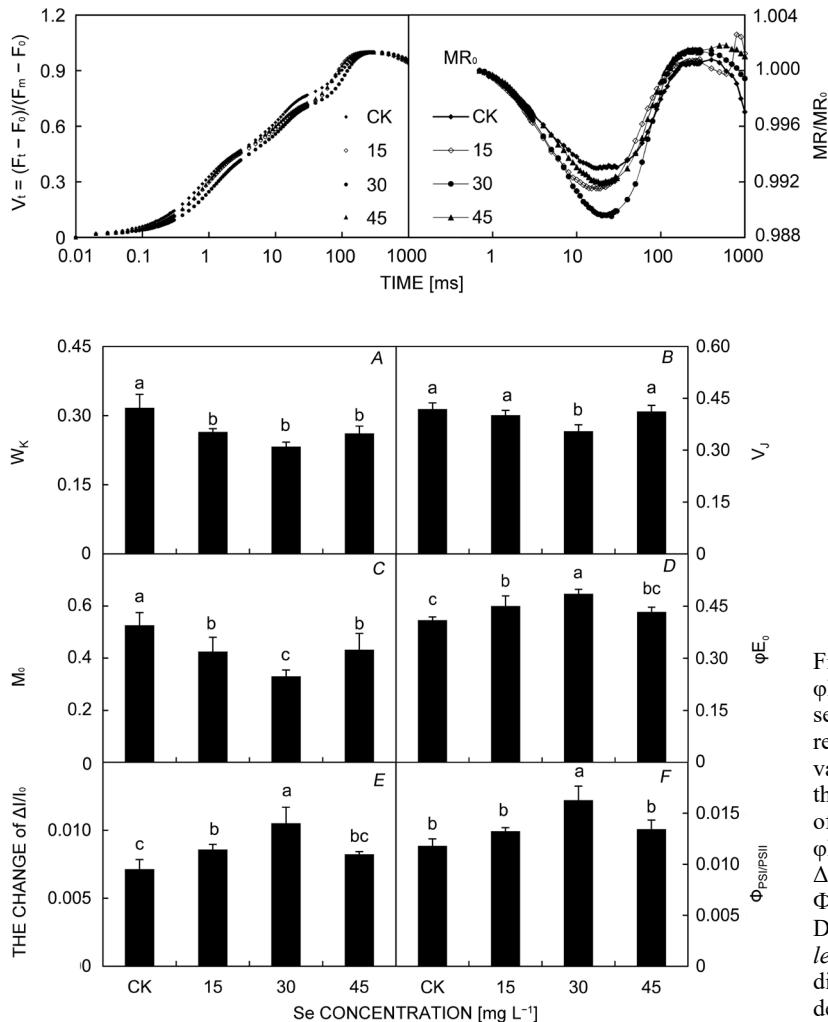


Fig. 3. The OJIP curve (A) and 820-nm reflection absorption curve (B) of tree peony under different selenite treatments.  $V_t$  – the relative variable fluorescence ( $V_t$ ) at any time; O, K, J, I, and P – different phases in the OJIP curve.  $MR/MR_0$  – the 820-nm modulated reflection curve; MR – the modulated reflection at different time points;  $MR_0$  – the MR value of far-red light irradiated at 0.7 ms.

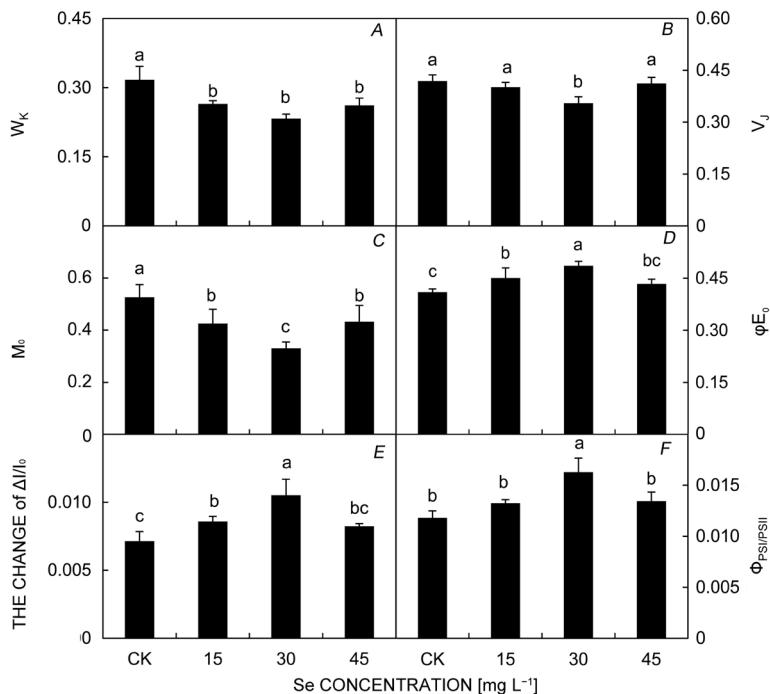


Fig. 4. The changes of  $W_K$  (A),  $V_J$  (B),  $M_0$  (C),  $\phi E_0$  (D),  $\Delta I/I_0$  (E), and  $\Phi_{PSI/PSII}$  (F) under different selenite treatments.  $W_K$  – the normalized relative variable fluorescence;  $V_J$  – the relative variable fluorescence intensity at the J step;  $M_0$  – the initial slope of the relative variable fluorescence of the relative rate at which  $Q_A$  is reduced;  $\phi E_0$  – the quantum yield for electron transport;  $\Delta I/I_0$  – the maximum redox activity of PSI;  $\Phi_{PSI/PSII}$  – the coordination between PSII and PSI. Data are means  $\pm$  SD ( $n = 12$ ). Different lowercase letters above the bars indicate a significant difference between treatments at  $P < 0.05$  as determined by a Duncan's multiple range test.

the trapped energy flux per cross-section ( $TR_0/CS_m$ ), the electron transport flux per cross-section ( $ET_0/CS_m$ ),  $F_v/F_m$ , and the performance index on absorption basis ( $PI_{abs}$ ) increased at first and then decreased with the increase of selenite concentration, reaching the maximum at 30 mg L<sup>-1</sup> selenite treatment. The dissipated energy flux per cross-section ( $DI_0/CS_m$ ) of all selenite treatments was significantly lower than those of the control.

$\Delta I/I_0$  and  $\Phi_{PSI/PSII}$  increased in all selenite treatments (Fig. 4). The  $\Delta I/I_0$  in 15 and 30 mg L<sup>-1</sup> selenite treatments were greatly higher than those in the control, except for the 45 mg L<sup>-1</sup> selenite treatment. The values of  $\Phi_{PSI/PSII}$  of all selenite treatments increased more than those in the control but were only greatly higher than those of the control under the 30 mg L<sup>-1</sup> selenite treatment. These two parameters all increased the most under the 30 mg L<sup>-1</sup> selenite treatment.

**Total flavonoids, total phenols, and secondary metabolites contents:** With the increase of selenite application concentration, the contents of total flavonoids (Fig. 5A) and total phenols (Fig. 5B) in MCR were increased first and then decreased. The contents of total flavonoids and total phenols in all selenite treatments were

higher than the control treatment, showing a significant difference in 15–30 mg L<sup>-1</sup> selenite treatments. It is worth mentioning that the contents of total flavonoids and total polyphenols in the 30 mg L<sup>-1</sup> selenite treatment increased more than those of other selenite treatments.

The changes in gallic acid (Fig. 5C) and paeonol (Fig. 5H) contents under different selenite treatments were similar, showing a trend of rising at first and then decreasing. These two parameters increased in all selenite treatments. The difference was that the content of gallic acid in the 30 mg L<sup>-1</sup> selenite treatment was significantly higher than that in the control, while the content of paeonol in the 15–30 mg L<sup>-1</sup> selenite treatments was significantly higher than that in the control. The change trend of catechin content (Fig. 5D) was similar to that of paeoniflorin content (Fig. 5F). The two parameters of all selenite treatments were significantly higher than those of the control, and those of 30–45 mg L<sup>-1</sup> selenite treatments were significantly higher than those of other treatments, but there was no significant difference between the two treatments. The content of albiflorin (Fig. 5E) in all selenite treatments was significantly higher than that in the control, and the maximum increase was observed in 30–45 mg L<sup>-1</sup> selenite treatments compared with the control group.

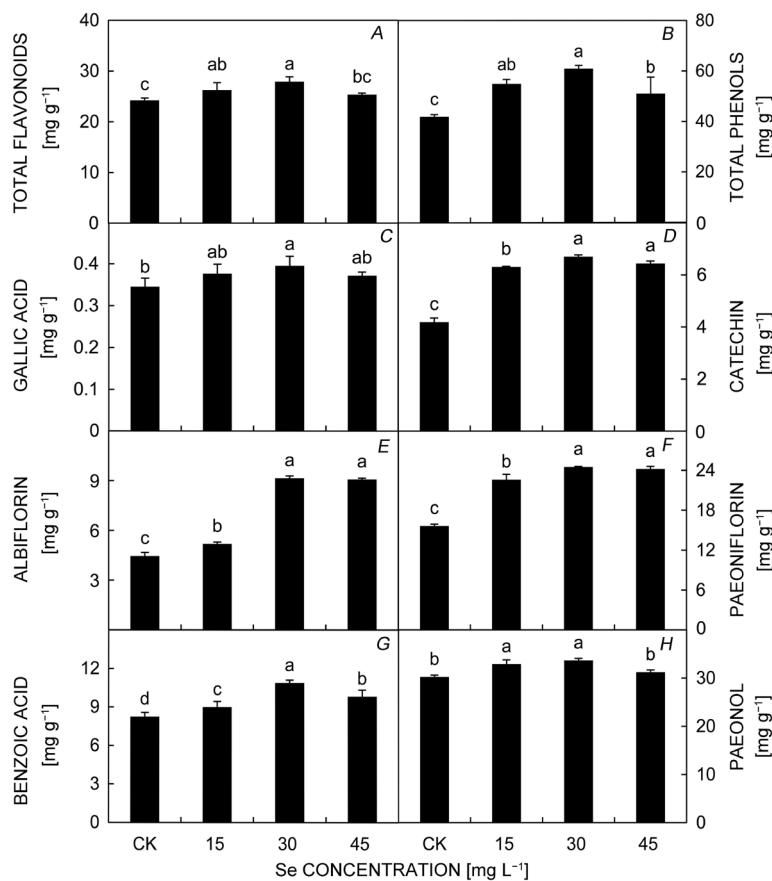


Fig. 5. Effect of different selenite treatments on the contents of total flavonoids (A), total polyphenols (B), gallic acid (C), catechin (D), albiflorin (E), paoniflorin (F), benzoic acid (G), and paeonol (H) in MCR of *Paeonia ostii*. MCR – Moutan cortex radicis. Data are means  $\pm$  SD ( $n = 3$ ). Different lowercase letters above the bars indicate a significant difference between treatments at  $P < 0.05$  as determined by a Duncan's multiple range test.

The content of benzoic acid (Fig. 5G) in all selenite treatments was significantly higher than that under the control, with the largest increase in the 30 mg L<sup>-1</sup> selenite treatment.

## Discussion

ROS, including H<sub>2</sub>O<sub>2</sub> and O<sub>2</sub><sup>·-</sup>, are produced in different cellular metabolic pathways, which can cause oxidative stress and damage plant cells at high concentrations (Hasanuzzaman *et al.* 2021). In this study, the control treatment had higher contents of REC and MDA, indicating the presence of electrolyte leakage and membrane lipid peroxidation in the control plants due to photooxidation damage, which was also confirmed by the higher accumulation of H<sub>2</sub>O<sub>2</sub> and O<sub>2</sub><sup>·-</sup> in the control leaves.

Selenite can be metabolized into a series of organo-selenium compounds in plants, and organic derivatives of selenium, such as selenocysteine and selenomethylthione, can be used as antioxidants to react directly with ROS (Rahmanto and Davies 2012). In addition, selenium-containing proteins/peptides, in particular, have the powerful ability to neutralize free radicals such as O<sub>2</sub><sup>·-</sup> and 2,2-biphenyl-1-picrylhydrazine (Zhu *et al.* 2019, Zhang *et al.* 2020). In this study, exogenous sodium selenite treatment obviously reduced the ROS content in *P. ostii*, which might be because sodium selenite application improved the synthesis of organic selenium compounds and selenium-containing proteins/peptides (Li *et al.*

2022b). On the other hand, previous studies have shown that selenite supplementation promoted the activities of enzymes such as GSH-Px, SOD, and POD (Chauhan *et al.* 2019). Consistent with previous studies, selenite pretreatment in this study increased SOD activity, which converts O<sub>2</sub><sup>·-</sup> to H<sub>2</sub>O<sub>2</sub>, providing first-line protection against ROS. Studies have shown that plants can accumulate osmotic regulators such as soluble proteins and proline to regulate cell osmotic potential and stabilize cell structure (Varghese *et al.* 2019). Of note, soluble sugar and proline are also involved in the detoxification of ROS in different organelles (Malik *et al.* 2022). In this study, selenite treatment greatly increased the contents of soluble protein, soluble sugar and proline in *P. ostii*, maintained the osmotic potential in cells, and cooperated with antioxidant enzymes to remove ROS in cells, thus reducing membrane lipid peroxidation, as indicated by a significant decrease of REC, MDA, and H<sub>2</sub>O<sub>2</sub> and O<sub>2</sub><sup>·-</sup> contents in *P. ostii*. Similarly, the application of sodium selenite increased the SOD and POD activities, decreased MDA content, and increased soluble protein, chlorophyll, and proline contents in *Raphanus sativus* (Hu *et al.* 2022). Low concentrations of sodium selenite significantly increased the proline and total soluble sugar contents, as well as SOD, CAT, POD, APX, and glutathione reductase (GR), and decreased the MDA and H<sub>2</sub>O<sub>2</sub> content in *Chenopodium quinoa* (Khalofah *et al.* 2021).

Appropriate selenite treatment increased the Chl content in *P. ostii* leaves, and similar results were also observed in

*C. quinoa* (Khalofah *et al.* 2021) and *R. sativus* (Hu *et al.* 2022). The increase in Chl content might occur because selenite application promoted the scavenging of ROS and protected Chl from photooxidation (Hawrylak-Nowak 2009). On the other hand, selenite application can increase the content of photosynthetic pigments by protecting chloroplast enzymes and promoting the biosynthesis of photosynthetic pigments (Chen *et al.* 2022).

In this study, the application of an appropriate amount of selenite induced an increase in  $P_N$  in *P. ostii* plants. As demonstrated in rice (Zhang *et al.* 2014), low concentrations of selenite had a positive effect on  $P_N$ , in support of our results. In this study,  $P_N$ ,  $g_s$ ,  $C_i$ , and  $E$  all increased after selenite treatments, suggesting that selenite could improve stomatal opening, which is conducive to  $\text{CO}_2$  exchange and water transpiration, thus increasing the photosynthetic capacity of *P. ostii* leaves. The results of this study are similar to those in the other plants after selenite application (Zhang *et al.* 2014, de Almeida *et al.* 2022).

$W_K$  reflects the injury of the oxygen-evolving complex (OEC) on the PSII donor side.  $V_J$  reflects the electron flow between quinone receptors  $Q_A$  and  $Q_B$ .  $M_0$  represents the reduction rate of the primary quinone receptor ( $Q_A$ ) and reflects the maximum closure rate of PSII reaction center.  $\phi E_0$  is the quantum of light energy absorbed by the reaction center for electron transfer (Zhang *et al.* 2017). In this study, the application of selenite (30 mg L<sup>-1</sup>) greatly increased the value of  $\phi E_0$  and decreased  $V_J$ , indicating the improvement of the electron transfer ability of the PSI, which was confirmed by the increase in ETR. The significant decrease of the  $W_K$  and  $M_0$  indicated that the application of selenite improved the function of OEC and integrity of the  $\text{Mn}_4\text{CaO}_5$  complex, and decreased the maximum closing rate of the PSII reaction center (Souza *et al.* 2019, Guo *et al.* 2020). ROS is formed by the leakage of electrons attacking  $\text{O}_2$  in the process of electron transport, and more ROS accumulation will damage the cell membrane and cause membrane peroxidation (Chalanika De Silva and Asaeda 2017). On the contrary, selenite treatment significantly improved the electron transport capacity in the photosystems, reduced electron leakage, decreased ROS contents, protected the stability of cell membrane and photosynthetic apparatus, and promoted the efficient operation of photosynthesis.

Compared with the control treatment, the ABS/RC,  $\text{TR}_0/\text{RC}$ ,  $\text{ET}_0/\text{RC}$ , and  $\text{DI}_0/\text{RC}$  in selenite treatment (30 mg L<sup>-1</sup>) were significantly reduced. The ABS/CS<sub>m</sub>,  $\text{TR}_0/\text{CS}_m$ , and  $\text{ET}_0/\text{CS}_m$  of the selenite treatment were significantly higher than those of the control treatment, while the  $\text{DI}_0/\text{CS}_m$  was significantly lower than that of the control. In particular, selenite treatment significantly increased RC/CS<sub>m</sub> of PSII. The above results indicated that selenite treatment could greatly reduce the energy charge pressure per reaction center of PSII by increasing the number of active reaction centers per cross-section (RC/CS<sub>m</sub>) of PSII, improve the efficiency of energy conversion and utilization, and reduce the occurrence of photoinhibition (Chang *et al.* 2023).

$F_v/F_m$  reflects the maximum photochemical efficiency of PSII under dark adaptation,  $F_v'/F_m'$  represents the maximum photochemical efficiency of PSII under light adaptation,  $\Phi_{\text{PSII}}$  reflects the actual photochemical efficiency of PSII, and ETR represents the rate of electron transfer,  $\text{PI}_{\text{abs}}$  is the performance index of photosynthetic apparatus, which can comprehensively reflect the performance of PSII (Strasser *et al.* 2010). In this study, the significant increase of RC/CS<sub>m</sub>,  $F_v/F_m$ ,  $F_v'/F_m'$ ,  $\Phi_{\text{PSII}}$ , ETR, and  $\text{PI}_{\text{abs}}$  of *P. ostii* showed that selenite treatment (30 mg L<sup>-1</sup>) significantly improved the maximum photochemical efficiency of PSII, the light energy-utilization efficiency and the photosynthetic performance in PSII. Similar results were also observed in *Billbergia zebrina* (Souza *et al.* 2019) and *Medicago sativa* after Se treatment (Wang *et al.* 2022).

The redox activity of PSI is represented by 820-nm modulated reflectance curve, and  $\text{MR}/\text{MR}_0$  reflects the ability of PSI reaction center to reduce terminal electron receptors (Yang *et al.* 2021).  $\Delta I/I_0$  is used to comprehensively evaluate the performance of PSI, and  $\Phi_{\text{PSI/PSII}}(\Delta I/I_0/\psi_0)$  represents the coordination between PSII and PSI (Zhang *et al.* 2021). In this study, the minimum value of  $\text{MR}/\text{MR}_0$ ,  $\Delta I/I_0$ , and  $\Phi_{\text{PSI/PSII}}$  decreased significantly after selenite treatment. These results showed that selenite treatment not only increased the activity of PSI reaction center, but also improved the mobility of electrons in both photosystems, thus significantly improving the coordination between PSII and PSI. Selenite treatment increased the content of photosynthetic pigments, the performance of PSII and PSI and the coordination between them, thus improving the photosynthetic capacity of leaves, which was also confirmed by the increase of  $P_N$  in *P. ostii*. In this study, selenite treatment increased photosynthesis, as well as soluble sugar and soluble protein content, finally provided sufficient carbohydrates for an increase in the total phenols and flavonoids and other substances.

Several studies showed phenols and flavonoid compounds can detoxify ROS, thereby protecting plant cells from photooxidation damage (Jucá *et al.* 2020, Wang *et al.* 2020b). In this study, the contents of total flavonoids, total phenols including gallic acid, catechins, and paeonol as well as benzoic acid, albiflorin and paeoniflorin were significantly increased by selenite treatment, which might contribute to scavenge ROS (Wang *et al.* 2017). Similar results were also reported in *Codonopsis lanceolata* (Zhu *et al.* 2017) and *B. juncea* (Li *et al.* 2023) after selenite treatment. In *C. lanceolata*, the contents of polysaccharides, total flavonoids, total saponins, proteins, total amino acids, and essential amino acids significantly increased at the level of the optimal selenium (Na<sub>2</sub>SeO<sub>3</sub>) treatment (1.0 mg L<sup>-1</sup>), which significantly improved the nutritive quality of *C. lanceolata*, while 2.0 mg L<sup>-1</sup> treatment reduced many nutritional indexes of *C. lanceolata* (Zhu *et al.* 2017). Application of selenium fertilizer (Na<sub>2</sub>SeO<sub>3</sub>) significantly increased the quality indexes of tomato fruits, such as total soluble solids, soluble sugar, titratable acid, sugar-acid ratio, vitamin C and lycopene, but had no significant effect on nitrate content and fruit hardness (Xu

*et al.* 2022). In this study, the 30 mg L<sup>-1</sup> selenite treatment significantly increased the contents of total flavonoids, total phenols, and six medicinal components in MCR, while the content of total flavonoids, total phenols, benzoic acid, and paeonol at 45 mg L<sup>-1</sup> of selenium showed a clear decrease. This suggests that the effect of selenium on quality improvement may vary depending on plant species, type of selenium fertilizer, concentration of Se, and method of fertilization.

Flavonoids are the largest group of phenolic compounds, and as other phenolic compounds are derived from phenylalanine (Balasundram *et al.* 2006). Studies have shown that selenite treatment of leaves induced the expression of key genes (*CHS* and *PAL*) in phenylpropionic acid metabolism in *Ginkgo biloba* leaves, and the transcription factors *MYB1* and *MYB2* involved in flavonoid biosynthesis were also significantly upregulated (Li *et al.* 2019). In *Arachis hypogaea*, selenite application upregulated the gene expression and related enzyme activities of the phenylpropanoid biosynthesis cascade, and finally increased the content of phenylpropanoid compounds including phenolic acid, lignin, and total flavonoids (Wang *et al.* 2016). In this study, the increase of flavonoids and phenols might be attributed to the upregulation of genes related to phenylpropionic acid metabolism induced by selenite treatment. Our results were consistent with previous studies on *C. lanceolata* (Zhu *et al.* 2017) and *Flammulina velutipes* (Dong *et al.* 2021) after sodium selenite application.

In the current study, selenite treatment also led to the enhancement of two monoterpenes constituents (paeoniflorin and albiflorin) in *P. ostii*. Recently, it has been found that selenite treatment can increase the content of monoterpenes (geranyl acetate, geranial, geraniol, nerol, and z-citral) (Azimi *et al.* 2021) and tetraterpenes (lycopen and β-carotene) (Morales-Espinoza *et al.* 2019) under non-stress conditions. Selenite treatment can affect transcription factors related to endogenous hormone regulation, and then regulate the expression of genes related to terpenoid synthesis, thereby promoting the accumulation of terpenoids, similar results were also found in *Ginkgo biloba* (Li *et al.* 2022a) treated with selenite.

Compared to soil application of selenium, foliar application is the most effective, safe, and economical method, and the efficiency of foliar application exceeds the efficiency of soil application by up to eight times (Poggi *et al.* 2000, Sattar *et al.* 2019). Xu *et al.* (2022) also found that Na<sub>2</sub>SeO<sub>3</sub> had a more significant promoting effect on fruit quality variables than Na<sub>2</sub>SeO<sub>4</sub> in tomato. Importantly, foliar application of selenite could greatly enhance the synthesis of organic selenium beneficial to humans and animals compared to selenate (Wang *et al.* 2020a). In this study, spraying sodium selenite significantly improved the antioxidant and photosynthetic capacity of medicinal *P. suffruticosa*, as well as the medicinal components in MCR, which has important application value in the cultivation of medicinal *P. suffruticosa*.

**Conclusions:** In this study, all selenite treatment had a positive effect on *P. ostii*, and the 30 mg L<sup>-1</sup> selenite

treatment was proved as the best. The 30 mg L<sup>-1</sup> selenite treatment significantly increased the antioxidant capacity of *P. ostii* by increasing the activity of SOD enzymes and nonenzymatic active substances, such as carotenoids, soluble sugars, total polyphenols, and total flavonoids, greatly reduced the ROS contents, and enhanced the protection of photosynthetic apparatus. Moreover, selenite treatment significantly increased the pigment contents, enhanced the activity and coordination of PSII and PSI, greatly improved the photosynthetic capacity, and further increased the contents of paeonol, paeoniflorin, and other secondary metabolites. The results showed that selenite increased the content of medicinal components of *P. ostii* by improving the antioxidant system, osmotic adjustment substance content and photosynthetic capacity of *P. ostii*. This study provided important information for revealing the influence of exogenous selenite on the physiology and medicinal components accumulation of *P. ostii* and has important application value in the efficient cultivation of medicinal tree peony.

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## Appendix 1. JIP parameters analysis.

Fluorescence parameters	Description
$W_K = (F_K - F_0)/(F_J - F_0)$	Normalized relative variable fluorescence
$V_J = (F_J - F_0)/(F_m - F_0)$	Relative variable fluorescence intensity at the J step
$M_0 = 4(F_{300\mu s} - F_0)/(F_m - F_0)$	Initial slope of the relative variable fluorescence of the relative rate at which Q <sub>A</sub> is reduced
$\phi E_0 = ET_0/ABS = [1 - (F_0/F_m)]\psi_0$	Quantum yield for electron transport
$ABS/RC = M_0(1/V_J)(1/\phi P_0)$	Absorption flux per reaction center
$TR_0/RC = M_0(1/V_J)$	Trapped energy flux per reaction center
$ET_0/RC = M_0(1/V_J)\psi E_0$	Electron transport flux per reaction center
$DI_0/RC = (ABS/RC) - (TR_0/RC)$	Dissipated energy flux per reaction center
$RC/CS_m = \phi P_0(V_J/M_0)(ABS/CS_m)$	Density of RCs per excited cross-section
$ABS/CS_m \approx F_m$	Absorbed energy flux per cross-section
$TR_0/CS_m = \phi P_0(ABS/CS_m)$	Trapped energy flux per cross-section
$ET_0/CS_m = \phi E_0(ABS/CS_m)$	Electron transport flux per cross-section
$DI_0/CS_m = ABS/CS_m - TR_0/CS_m$	Dissipated energy flux per cross-section
$F_v/F_m'$	Efficiency of excitation capture of open PSII center
$F_v/F_m$	Maximal quantum yield of PSII photochemistry
$PI_{abs} = (RC/ABS)[\phi P_0/(1 - \phi P_0)][\psi_0/(1 - \psi_0)]$	Performance index on absorption basis